

# BiSX and BiSX/g-C<sub>3</sub>N<sub>4</sub> (X = Cl, Br, I): Synthesis, Characterization, Photocatalysis of Carbon Dioxide, and Crystal Violet Dye

科學教育與應用學系 四年級 曾智暄

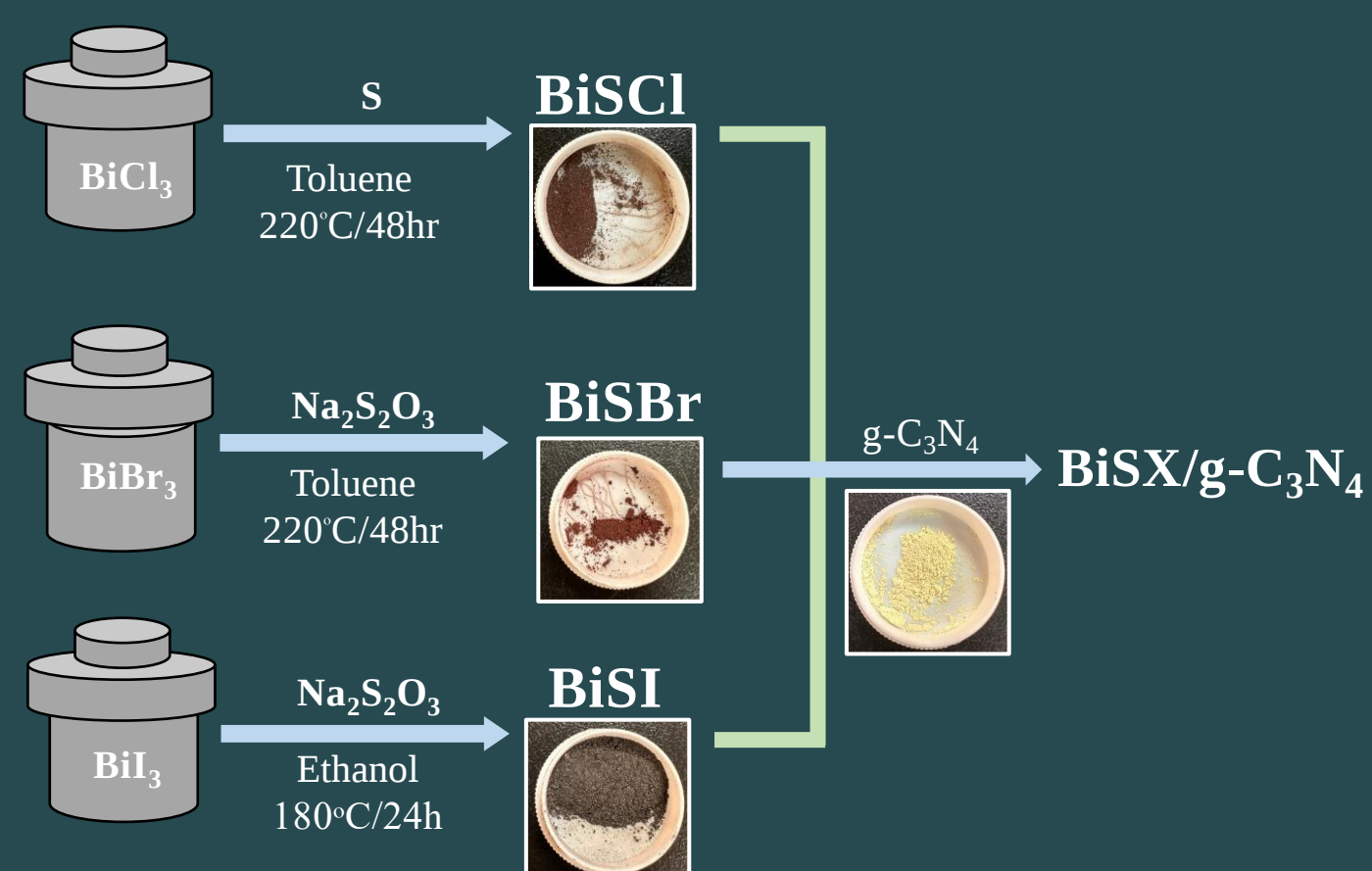
指導教授 黃旭村 教授兼系主任,陳錦章 卓越講座教授兼理學院院長

## Abstract

Using a simple hydrothermal method, we successfully synthesized a series of BiSX compounds, belonging to the group of V–VI–VII compounds, which includes BiSbCl, BiSbBr, and BiSI. These compounds have been under investigation since the last century due to their potential for optoelectronic applications. The binary composite photocatalysts BiSX/g-C<sub>3</sub>N<sub>4</sub> were prepared by mixing varying weights of g-C<sub>3</sub>N<sub>4</sub> and heating the mixture at 100°C for 4 hours. The resulting products underwent characterization using XRD, FE-TEM, FT-IR, SEM-EDS, DR-UV, BET, PL, EPR, and HR-XPS techniques. Discussion on the photocatalytic efficiency of BiSX and BiSX/g-C<sub>3</sub>N<sub>4</sub> demonstrates significant effects, as these catalysts were utilized for converting CO<sub>2</sub> into hydrocarbons and for the photocatalytic degradation of the organic pollutant, crystal violet (CV), suggesting their promising potential in mitigating environmental pollution.

## Results and Discussion

### Catalysts Preparation



### XRD

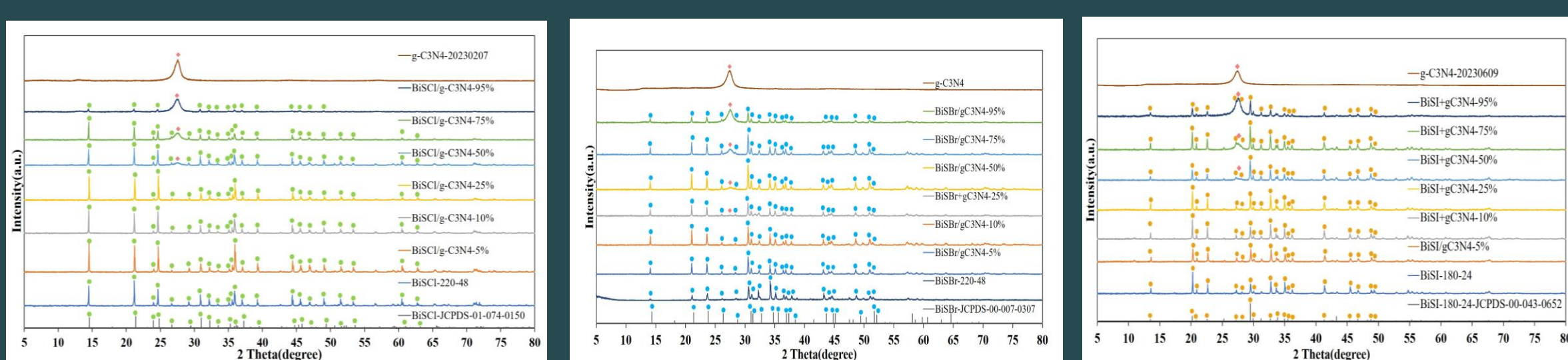


Fig.1 XRD patterns of BiSX (X = Cl, Br, I) and BiSX/g-C<sub>3</sub>N<sub>4</sub>.

### FE-SEM-EDS

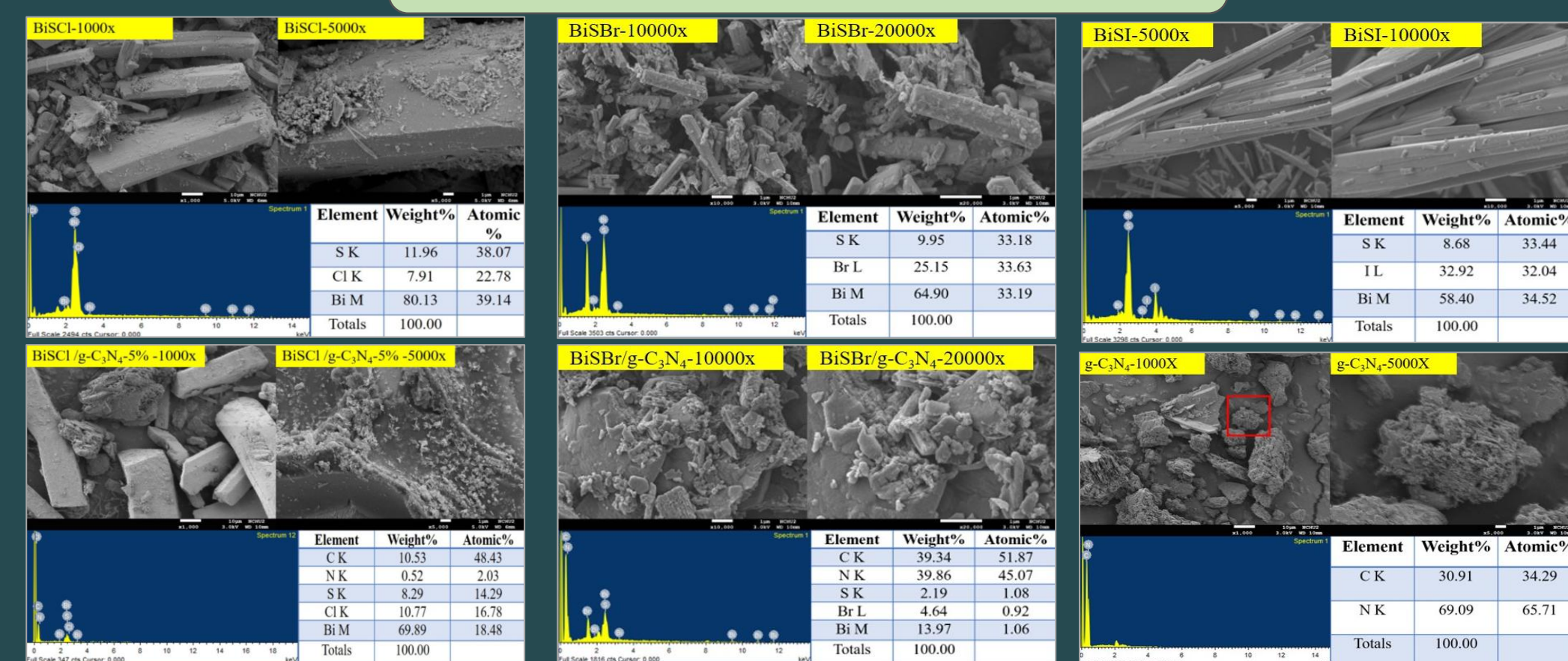


Fig.2 FE-SEM images and EDS of BiSX (X = Cl, Br, I) and BiSX/g-C<sub>3</sub>N<sub>4</sub>.

### FE-TEM-EDS

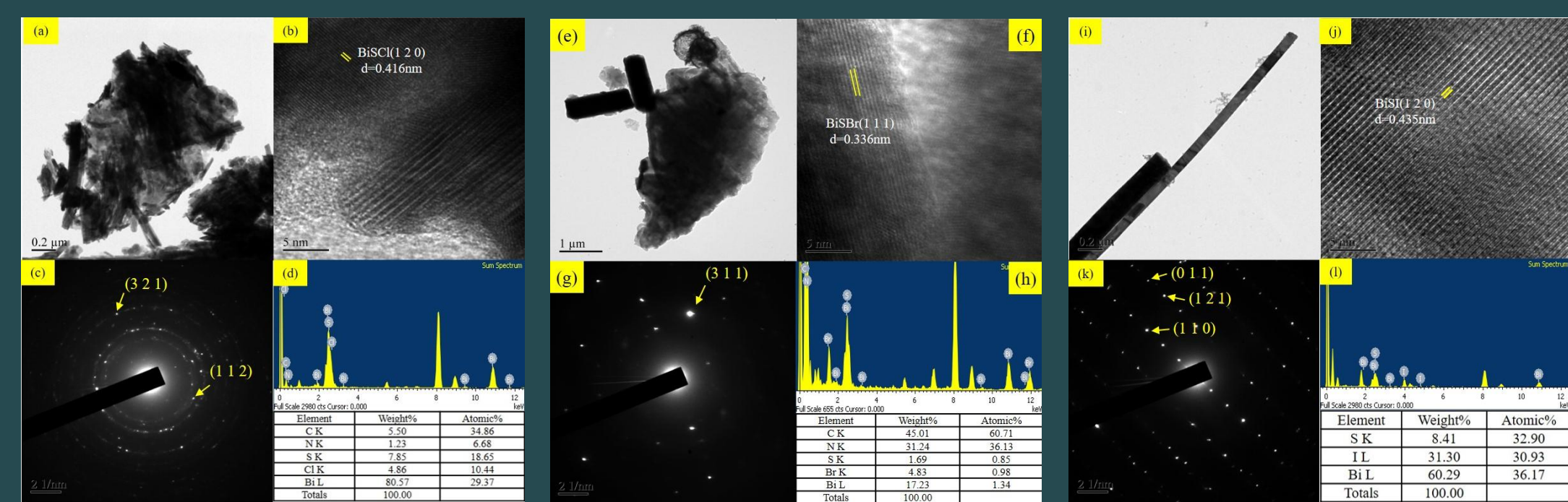


Fig.3 (a) FE-TEM image, (b) HR-TEM image, (c) SAED and (d) EDS of BiSbCl/g-C<sub>3</sub>N<sub>4</sub>-5wt% sample, (e) FE-TEM image, (f) HR-TEM image, (g) SAED and (h) EDS of BiSbBr/g-C<sub>3</sub>N<sub>4</sub>-95wt% sample, (i) FE-TEM image, (j) HR-TEM image, (k) SAED and (l) EDS of BiSI sample.

### HR-XPS

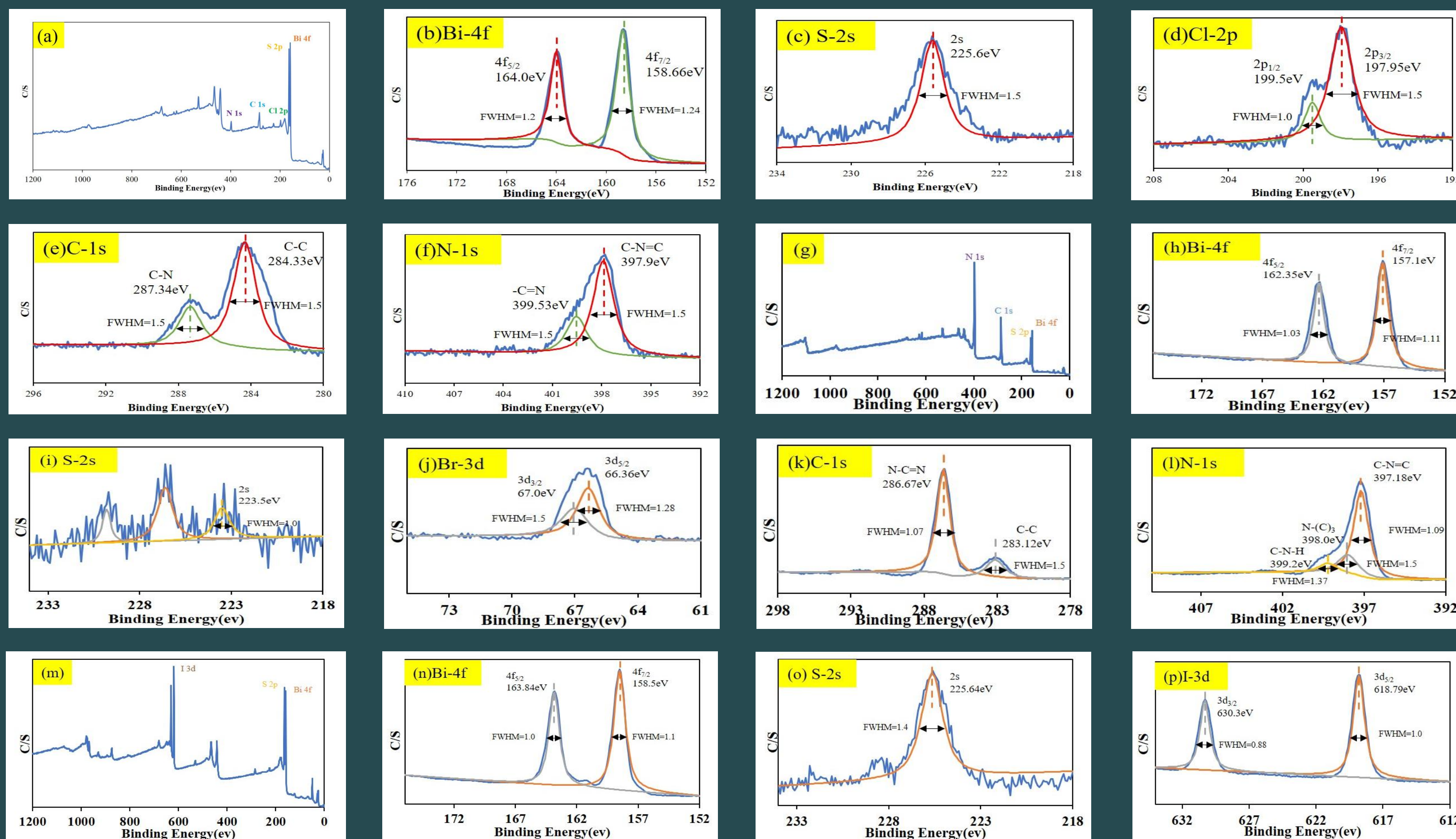


Figure 4. XPS spectra of the as-prepared BiSX/g-C<sub>3</sub>N<sub>4</sub> samples: (a) survey XPS spectrum; high-resolution XPS spectra of (b) Bi 4f, (c) S 2s, (d) Cl 2p, (e) C 1s and (f) N 1s of the BiSbCl/g-C<sub>3</sub>N<sub>4</sub>-5wt%; (g) survey XPS spectrum; high-resolution XPS spectra of (h) Bi 4f, (i) S 2s, (j) Br 3d, (k) C 1s and (l) N 1s of the BiSbBr-g-C<sub>3</sub>N<sub>4</sub>-95wt%; (m) survey XPS spectrum; high-resolution XPS spectra of (n) Bi 4f, (o) S 2s, (p) I 3d of the BiSI.

### Photocatalytic Degradation of CV

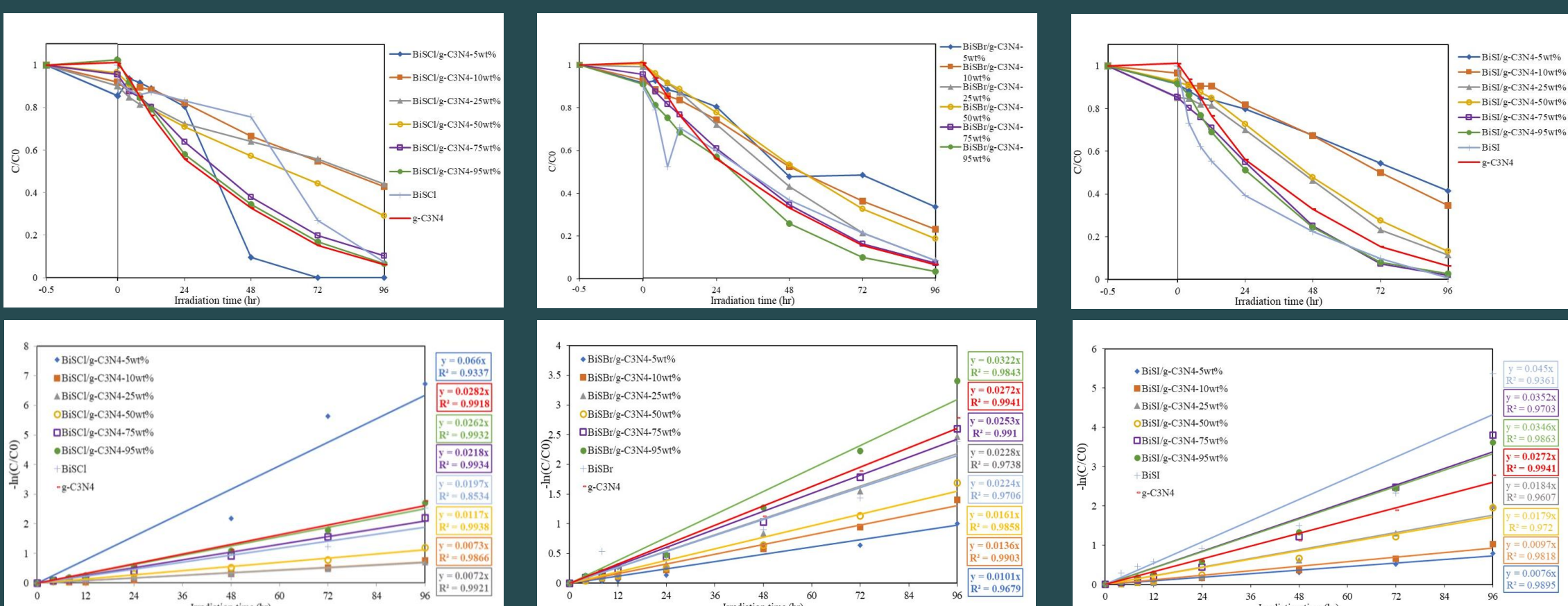


Fig.5 Photodegradation of CV as a function of irradiation time over different BiSX/g-C<sub>3</sub>N<sub>4</sub> photocatalysts.

### Photoreduction of CO<sub>2</sub>

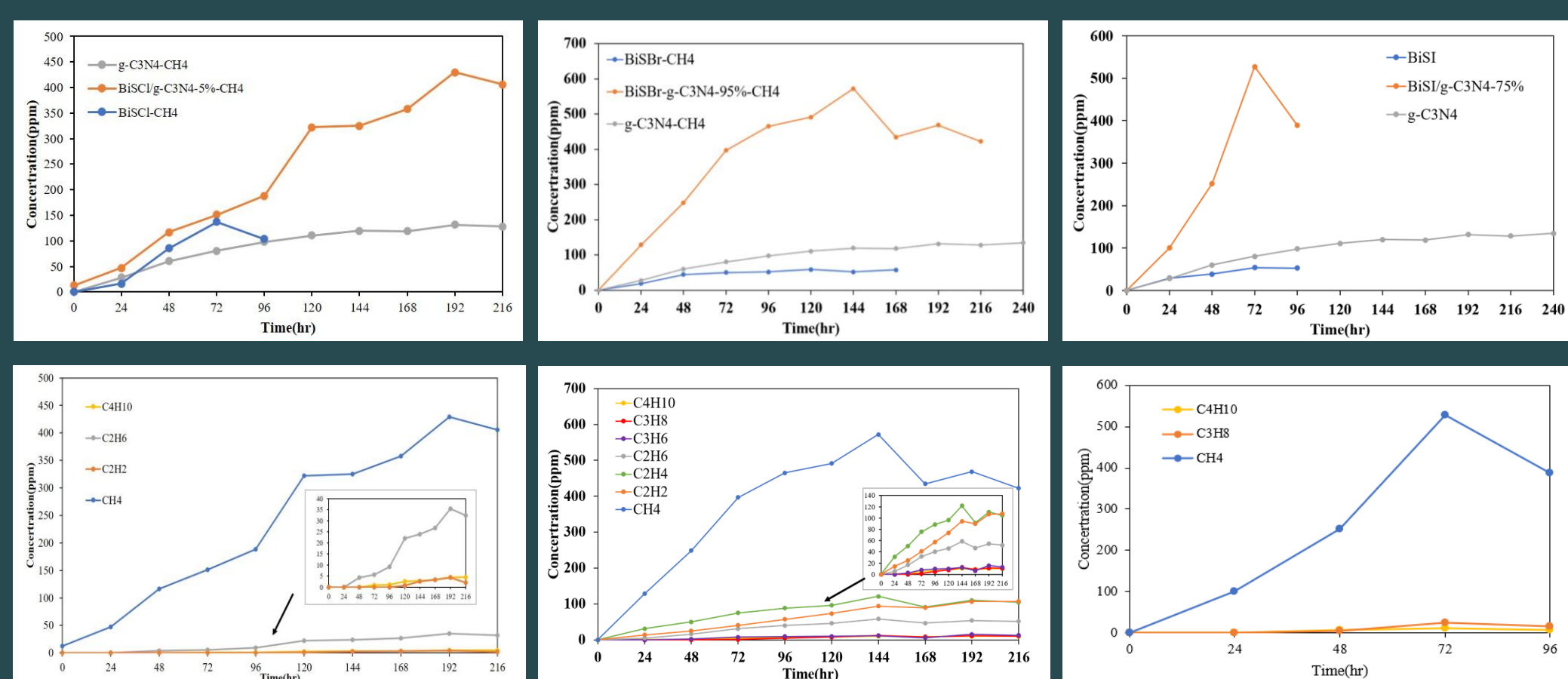


Fig.6 Photocatalytic reduction of CO<sub>2</sub> by BiSX/g-C<sub>3</sub>N<sub>4</sub>.

### BET

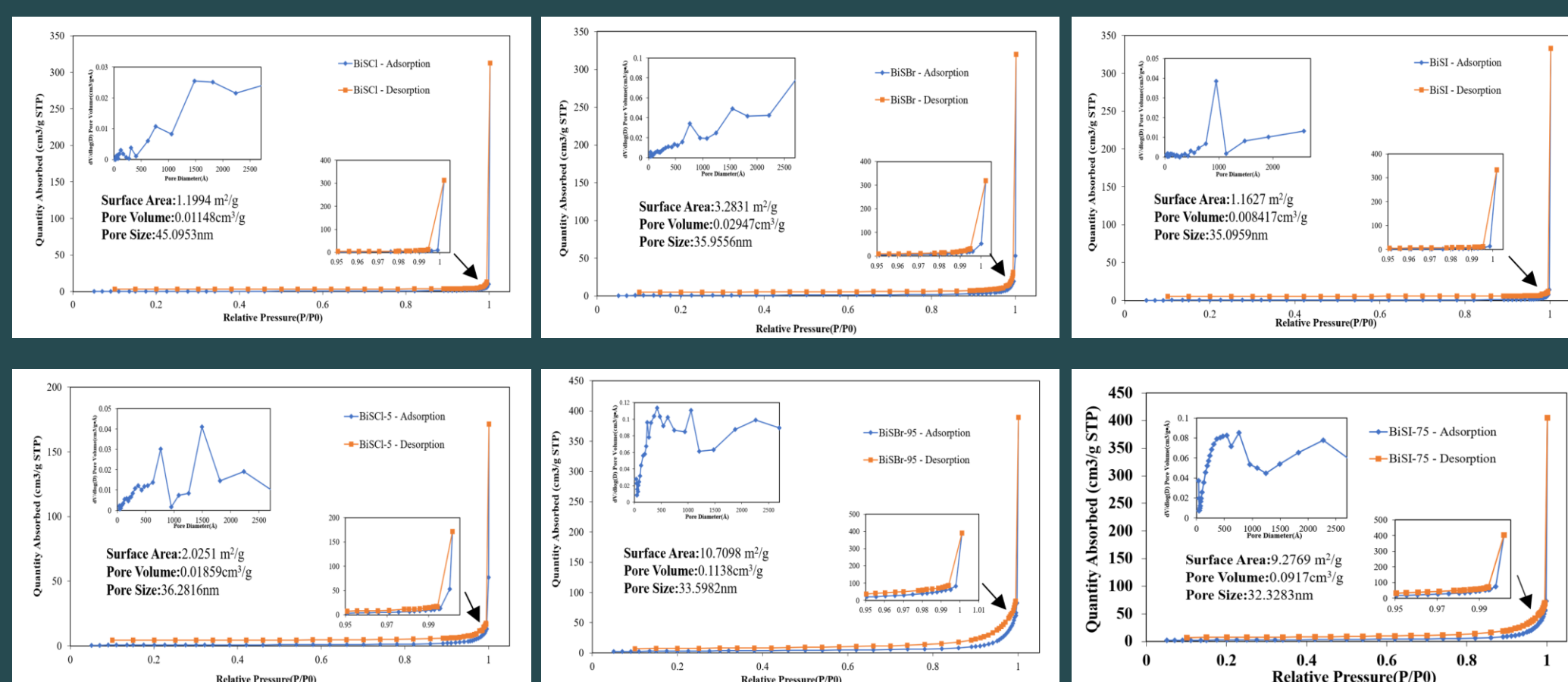


Fig.7 N<sub>2</sub> adsorption isotherm and pore size distribution of BiSX, BiSX/g-C<sub>3</sub>N<sub>4</sub>.

### Mechanism

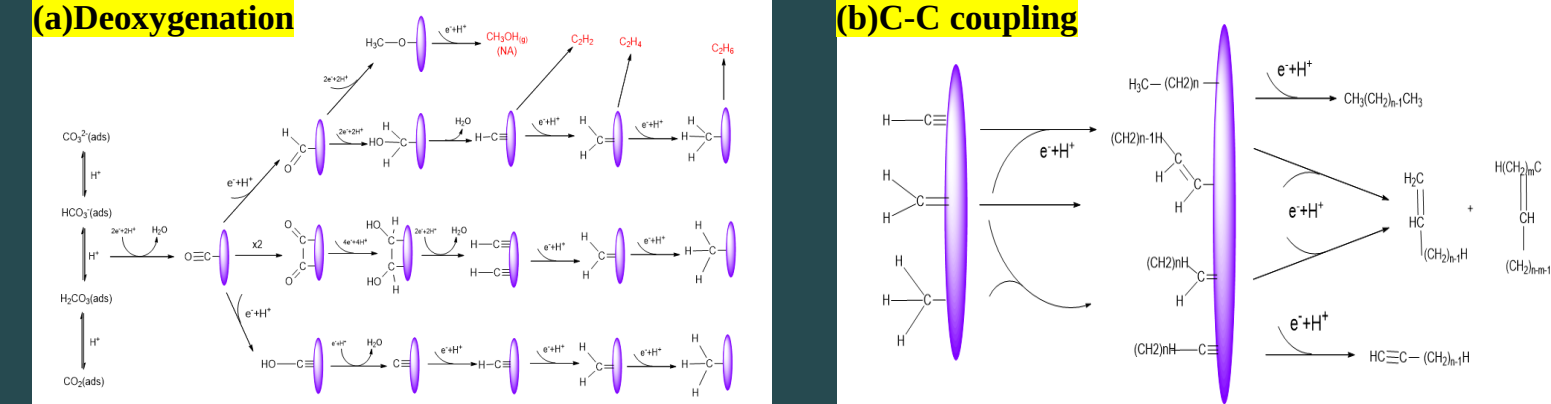


Figure 11. Proposed mechanisms of (a) deoxygenation and (b) C-C coupling on photoreduction CO<sub>2</sub> with BiSX/g-C<sub>3</sub>N<sub>4</sub> photocatalysts.

### Scheme

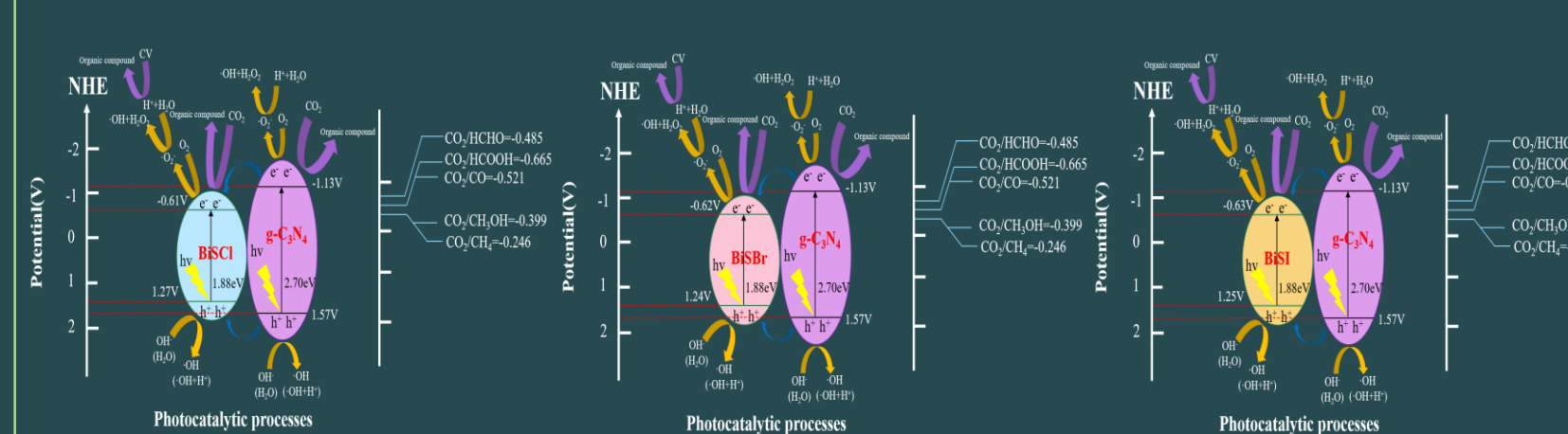


Fig.12 Band structure diagram and possible charge separation processes of BiSX/g-C<sub>3</sub>N<sub>4</sub> for the photoreduction of CO<sub>2</sub>.

### Summary

The utilization of BiSX and BiSX/g-C<sub>3</sub>N<sub>4</sub> for CO<sub>2</sub> reduction and photocatalytic degradation of organic pollutants, such as crystal violet (CV), demonstrates significant effects, indicating their promising potential for reducing environmental pollution.

### UV-Vis-NIR

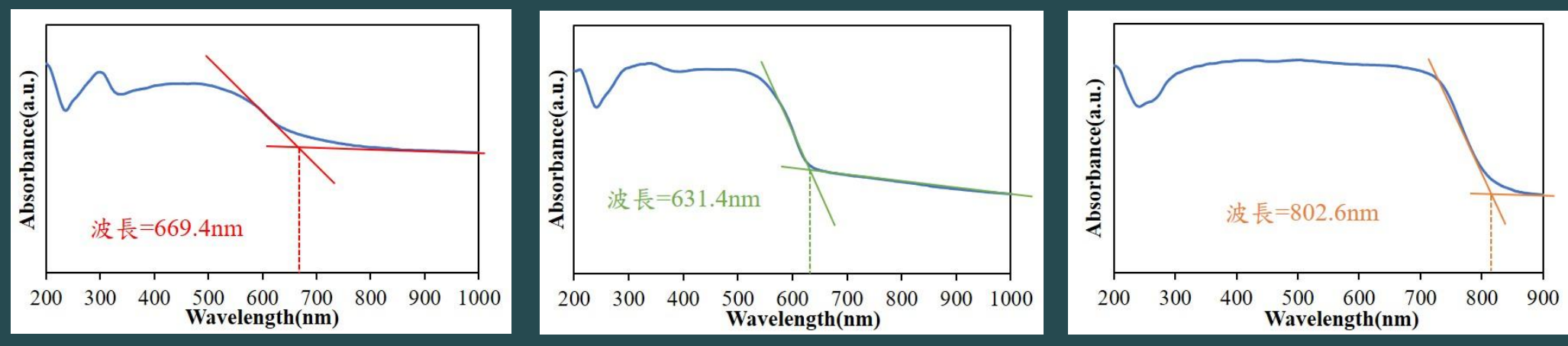


Fig.8 NIR pattern of BiSX.

### PL

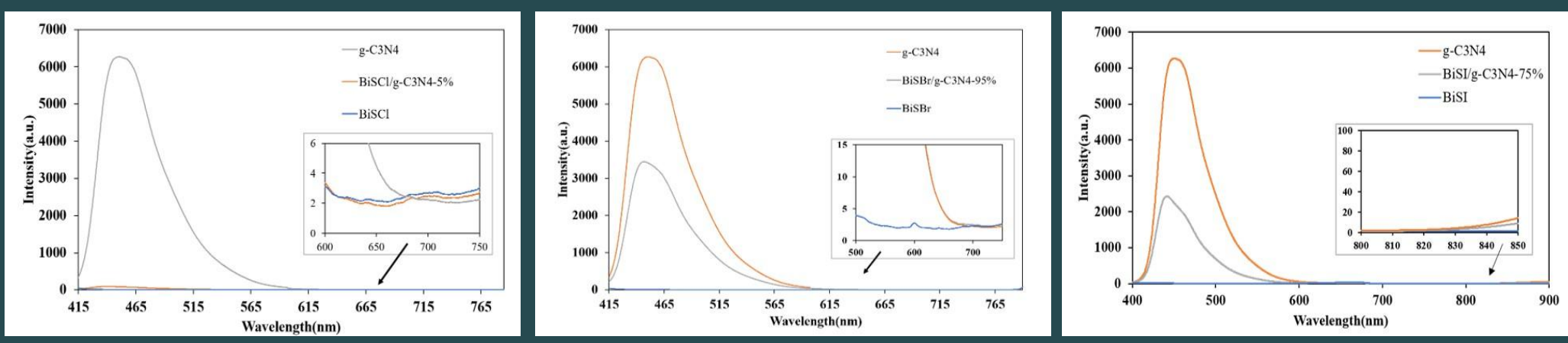


Fig.9 PL pattern of BiSX, BiSX/g-C<sub>3</sub>N<sub>4</sub> and g-C<sub>3</sub>N<sub>4</sub>.

### EPR

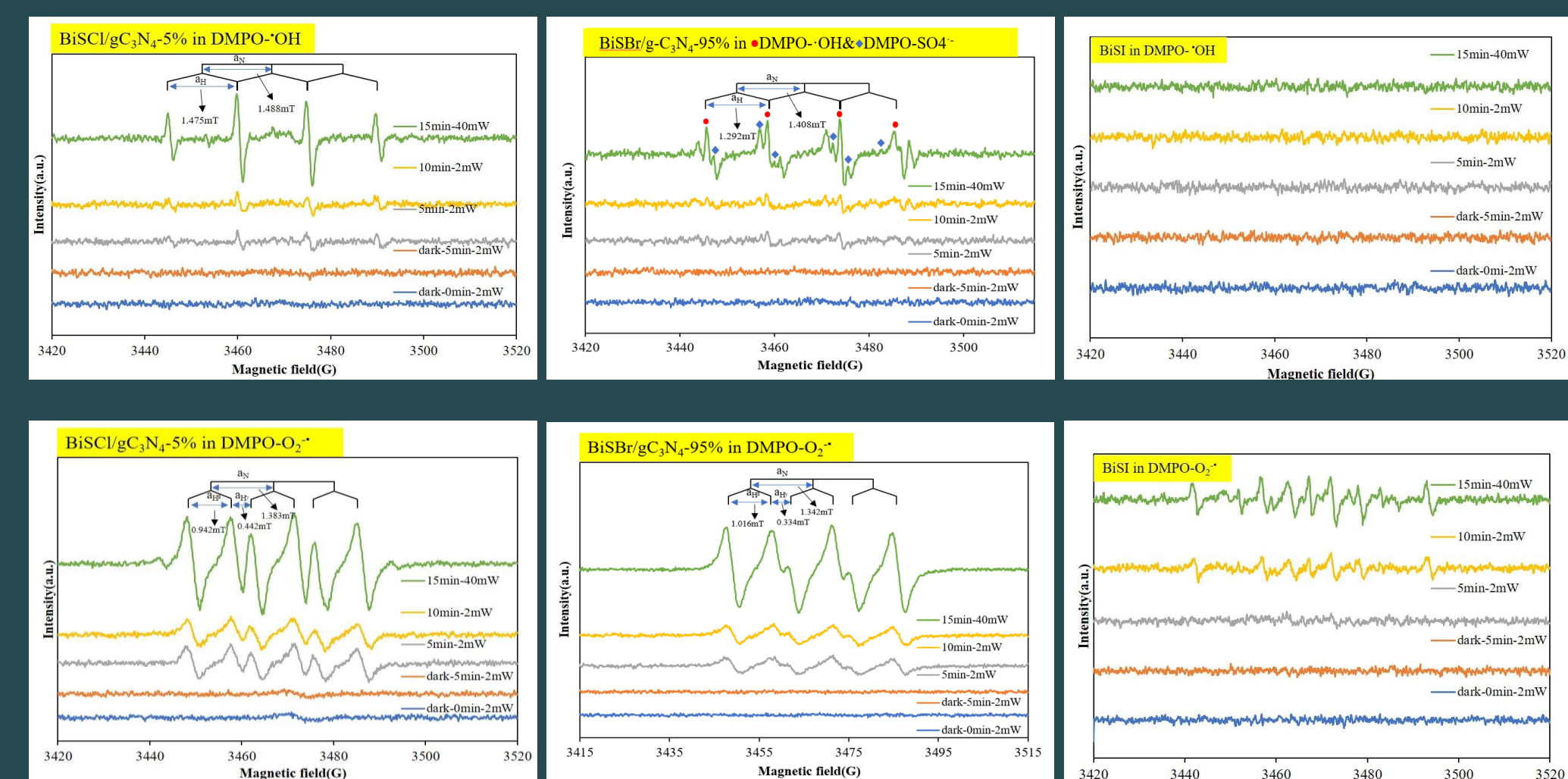


Fig.10 EPR pattern of BiSbCl/g-C<sub>3</sub>N<sub>4</sub>-5wt%, BiSbBr/g-C<sub>3</sub>N<sub>4</sub>-95wt% and BiSI.