

# Theoretical study of the photoelectron spectra of 1,3-Pentadiyne

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## Abstract

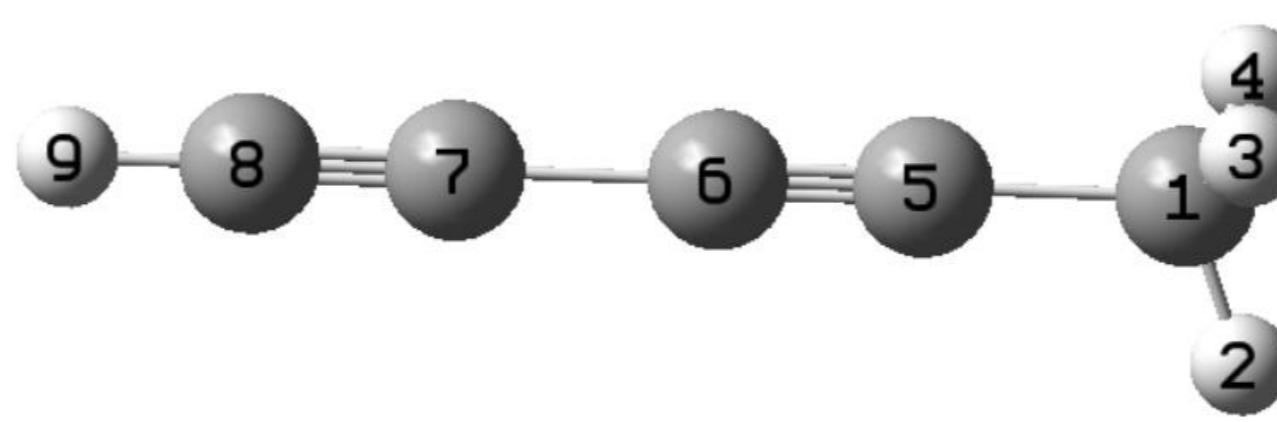
We used density functional theory B3LYP, PBE0,  $\omega$ B97XD, and APFD methods with the aug-cc-pVTZ basis set to calculate the equilibrium structure and vibration frequency of the 1, 3-pentadiyne molecule and cation. We also calculated the Frank–Condon factor using the harmonic oscillator model developed in our laboratory. We then simulated the photoelectron spectrum of the 1, 3-pentadiyne molecule when it stripped electrons to form cations and compared it with the experimental spectrum.

## Results and Discussion

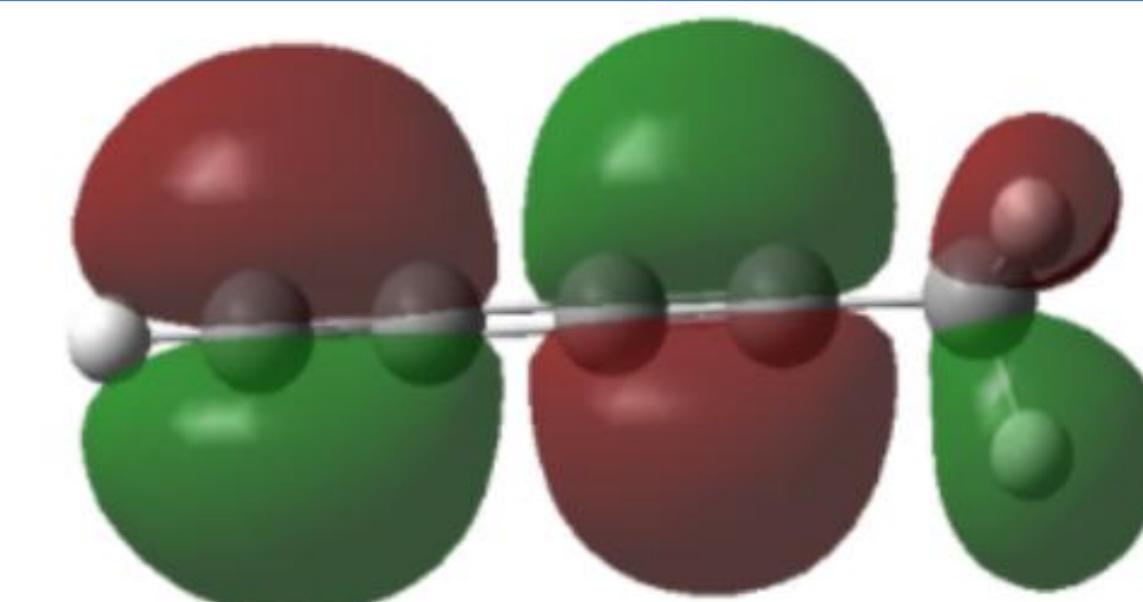
### Equilibrium Structures

**Table 1.** The equilibrium structures calculated by different methods.

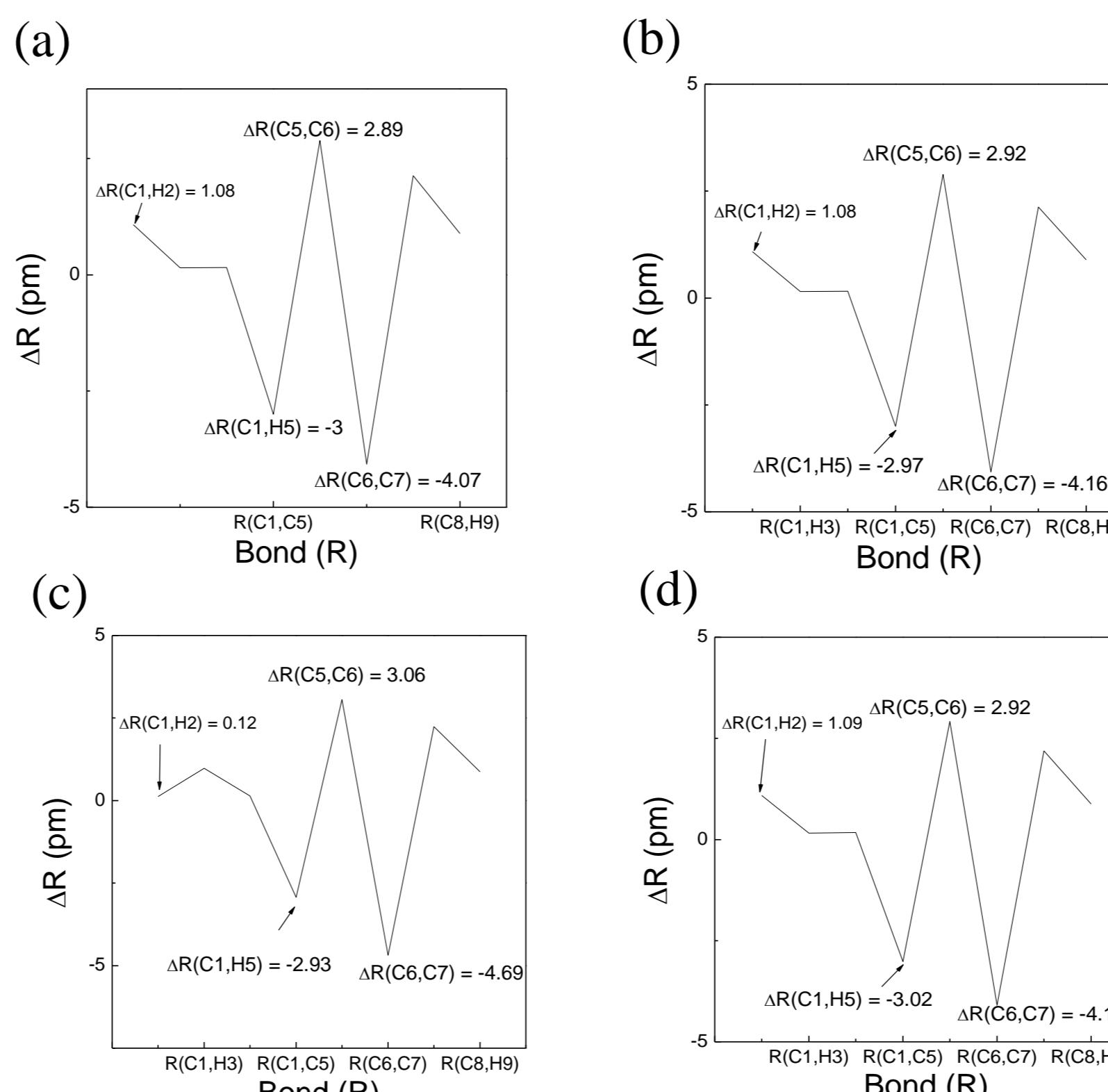
Parameter	Method							
	B3LYP		PBE0		$\omega$ B97XD		APFD	
X	X+	X	X+	X	X+	X	X+	
R(C1,H2)	109.16	110.24	109.24	110.32	109.00	109.12	109.27	110.36
R(C1,H3)	109.16	109.31	109.24	109.36	109.00	109.98	109.27	109.43
R(C1,H4)	109.16	109.32	109.24	109.43	109.00	109.14	109.27	109.44
R(C1,C5)	145.05	142.05	144.52	141.55	145.18	142.25	144.61	141.59
R(C5,C6)	120.69	123.58	120.69	123.61	120.12	123.18	120.69	123.61
R(C6,C7)	136.35	132.28	136.33	132.17	137.16	132.47	136.24	132.14
R(C7,C8)	120.54	122.67	120.52	122.71	120.00	122.24	120.56	122.75
R(C8,H9)	106.11	107.00	106.33	107.22	106.20	107.07	106.34	107.22
A(H2,C1,H3)	107.9	107.3	108.0	107.6	108.3	107.8	108.0	107.3
A(H2,C1,H4)	107.9	107.3	108.0	107.2	108.3	111.5	108.0	107.3
A(H2,C1,C5)	111.0	107.9	110.9	107.6	110.6	111.1	110.9	107.7
A(H3,C1,H4)	107.9	111.3	108.0	111.5	108.3	107.7	108.0	111.4
A(H3,C1,C5)	111.0	111.4	110.9	111.5	110.6	107.4	110.9	111.4
A(H4,C1,C5)	111.0	111.4	110.9	111.2	110.6	111.0	110.9	111.4



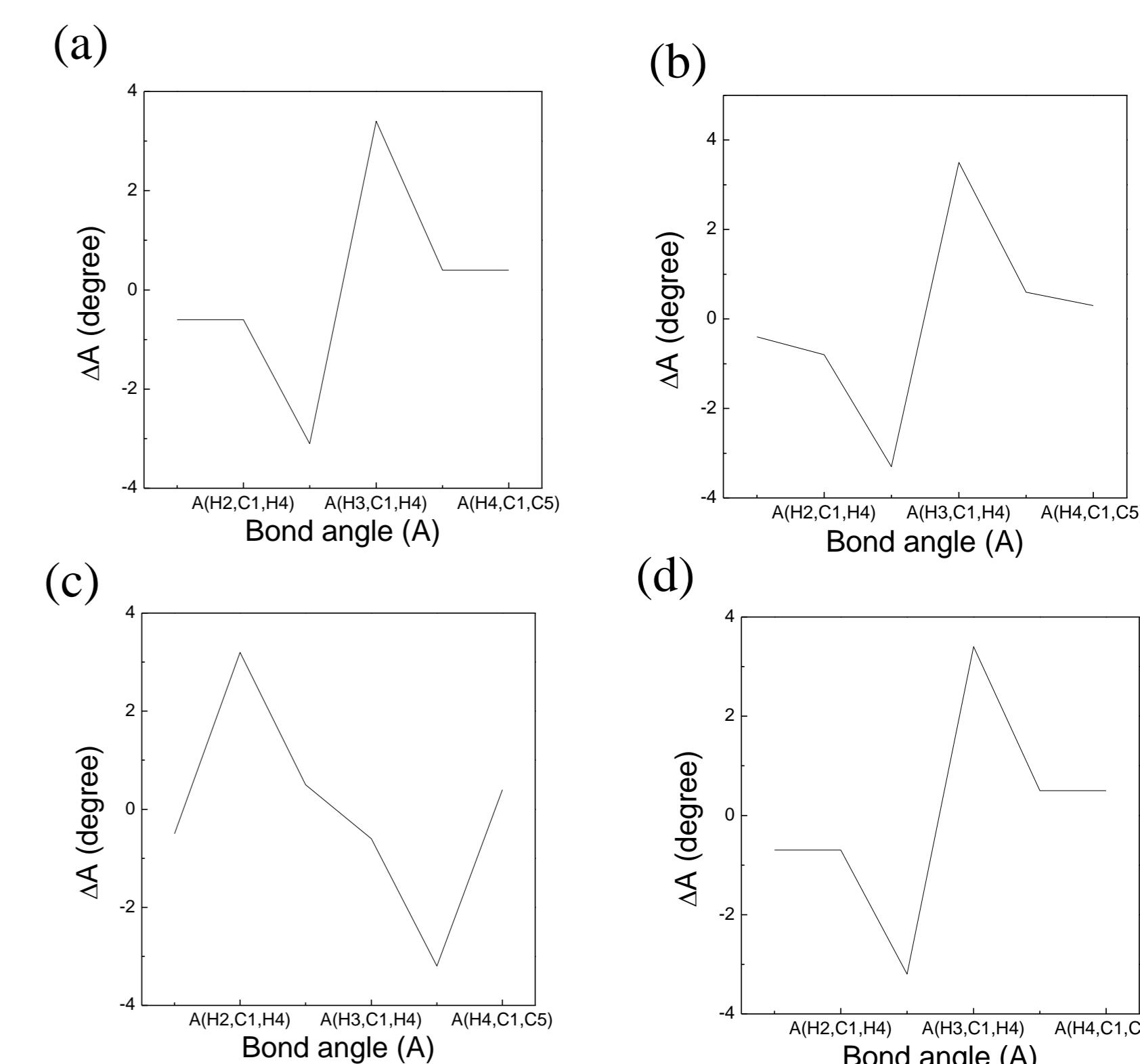
**Fig. 1.** The equilibrium structures of 1, 3-pentadiyne.



**Fig. 2.** The HOMO diagram of 1, 3-pentadiyne.



**Fig. 3.** The bond length variation diagram calculated by (a) B3LYP, (b) PBE0, (c)  $\omega$ B97XD, (d) APFD.



**Fig. 4.** The bond angle variation diagram calculated by (a) B3LYP, (b) PBE0, (c)  $\omega$ B97XD, (d) APFD.

### Vibrational Frequencies

**Table 2.** The harmonic vibrational frequency calculated by B3LYP, PBE0,  $\omega$ B97XD, and APFD /aug-cc-pVTZ.

mode	B3LYP		PBE0		$\omega$ B97XD		APFD	
	frequency	Symmetry species	frequency	Symmetry species	frequency	Symmetry species	frequency	Symmetry species
1	3468	A	3383	A	3480	A	3393	A
2	3078	A	3044	A	3112	A	3084	A
3	3078	A	3040	A	3112	A	3066	A
4	3022	A	2958	A	3042	A	2982	A
5	2343	A	2285	A	2371	A	2313	A
6	2165	A	1997	A	2184	A	2009	A
7	1473	A	1433	A	1462	A	1423	A
8	1473	A	1349	A	1462	A	1339	A
9	1415	A	1286	A	1402	A	1282	A
10	1179	A	1229	A	1198	A	1246	A
11	1050	A	955	A	1039	A	940	A
12	1050	A	756	A	1039	A	755	A
13	680	A	694	A	690	A	703	A
14	645	A	643	A	657	A	654	A
15	645	A	516	A	657	A	528	A
16	529	A	514	A	533	A	510	A
17	529	A	455	A	533	A	464	A
18	340	A	347	A	344	A	347	A
19	340	A	261	A	344	A	266	A
20	147	A	139	A	148	A	139	A
21	147	A	138	A	148	A	136	A

### Franck-Condon Factor

**Table 3.** The Franck-Condon factors calculated by B3LYP, PBE0,  $\omega$ B97XD, and APFD method.

HC≡C—C≡C—CH <sub>3</sub> → HC≡C—C≡C—CH <sub>3</sub> <sup>+</sup> + e <sup>-</sup>											
Method											
B3LYP		PBE0		$\omega$ B97XD		APFD					
ΔE	FCF	State	ΔE	FCF	State	ΔE	FCF	State	ΔE		
0	4.8954E-01	0 <sup>0</sup>	0	4.8276E-01	0 <sup>0</sup>	0	4.9582E-01	0 <sup>0</sup>	0	4.8060E-01	0 <sup>0</sup>
347	7.2356E-03	18 <sup>1</sup>	347	9.0544E-03	18 <sup>1</sup>	348	8.5035E-03	18 <sup>1</sup>	346	8.9461E-03	18 <sup>1</sup>
522	1.6487E-02	19 <sup>2</sup>	533	1.4196E-02	19 <sup>2</sup>	516	4.9702E-03	16 <sup>1</sup>	512	4.9610E-03	16 <sup>1</sup>
694	3.3554E-02	13 <sup>1</sup>	703	3.3672E-02	13 <sup>1</sup>	697	4.3055E-02	13 <sup>1</sup>	519	1.6185E-02	19 <sup>2</sup>
910	7.5663E-03	17 <sup>2</sup>	928	5.6807E-03	17 <sup>2</sup>	962	2.2618E-02	11 <sup>1</sup>	701	3.3185E-02	13 <sup>1</sup>
955	2.3680E-02	11 <sup>1</sup>	940	2.5108E-02	11 <sup>1</sup>	1279	1.3250E-02	15 <sup>2</sup>	916	6.3144E-03	17 <sup>2</sup>
1433	8.7232E-03	7 <sup>1</sup>	1423	8.3584E-03	7 <sup>1</sup>	1443	8.8871E-03	7 <sup>1</sup>	937	2.4415E-02	11 <sup>1</sup>
2285	1.8106E-01	5 <sup>1</sup>	2313	1.8717E-01	5 <sup>1</sup>	2343	2.1745E-01	5 <sup>1</sup>	1422	8.1276E-03	7 <sup>1</sup>
2807	6.0978E-03	5 <sup>1</sup> 19 <sup>2</sup>	2846	5.5041E-03	5 <sup>1</sup> 19 <sup>2</sup>	3040	2.3843E-02	5 <sup>1</sup> 13 <sup>1</sup>	2307	1.8396E-01	5 <sup>1</sup>
2979	1.6221E-02	5 <sup>1</sup> 13 <sup>1</sup>	3017	1.7035E-02	5 <sup>1</sup> 13 <sup>1</sup>	3305	9.0679E-03	5 <sup>1</sup> 11 <sup>1</sup>	2826	6.1951E-03	5 <sup>1</sup> 19 <sup>2</sup>
3240	8.0505E-03	5 <sup>1</sup> 11 <sup>1</sup>	3253	8.9363E-03	5 <sup>1</sup> 11 <sup>1</sup>	3622	5.8112E-03	5 <sup>1</sup> 15 <sup>2</sup>	3008	1.6583E-02	5 <sup>1&lt;/</sup>